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(Numericals) |



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QUESTION 1.1 TO 1.26), CLASS -12

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Numericals Chemistry Chapter Solid State

(v) 13-15 group compounds : When the

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**Solid** state materials are produced by combination of elements of groups 13 and 15, the compounds thus obtained are called 13-15 compounds.

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The post Solid State Chemistry  
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Questions related to the topic Que 1. A solid has a cubic structure in which X atoms are located at the corners of the cube, Y atoms are at the cube centers and O atoms are at the edge centers. What is the formula of the compound?

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State  
length & Nearest neighbor distance (c)  
Edge length & radius (d) Density of unit  
cell.

Numericals of Solid State , Class 12th  
Chemistry | Exam ...

Hello friends, this video contains all the  
solved ncert numericals of the chemistry

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Solid state numericals

(v) 13-15 group compounds : When the solid state materials are produced by combination of elements of groups 13 and 15, the compounds thus obtained are



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## Chemistry Chapter Solid

State called 13-15 compounds. For example, InSb, AlP, GaAs, etc. 12-16 group compounds : Combination of elements of groups 12 and 16 yield some solid compounds which are referred to as 12-16 compounds. For example, ZnS, CdS, CdSe, HgTe, etc.

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## Chemistry Chapter Solid

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The Solid State Solids Solids are the chemical substances which are characterised by definite shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules

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States) are closely packed and held together by strong interparticle forces  
Types of Solids

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State Chemistry Chapter 1 To discover more about this chapter, browse through these Solid State Class 12 Notes from BYJU ' S.

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deal with arrangements of particles in solids that result in several types of structure in solids. These differences in the structural unit also result in different properties of each Solid-State.

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Physics Chemistry Biology. We hope the  
NCERT Exemplar Class 12 Chemistry  
Chapter 1 Solid State help you. If you  
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Solid State, drop a comment below and  
we will get back to you at the earliest.

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## Chemistry Chapter Solid State

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Ans: The main concepts under this chapter are: Characteristics and properties of the solid-state Amorphous and Crystalline solids Classification of crystalline solids Crystal Lattices and unit

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State  
cells Unit cells Types of unit cells Crystal  
lattices Characteristics of a crystal lattice  
Number of atoms ...

Class 12 Chemistry Revision Notes for  
Chapter 1 - The ...

Numericals Chemistry Chapter Solid State  
(v) 13-15 group compounds : When the



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**Solid** state materials are produced by combination of elements of groups 13 and 15, the compounds thus obtained are called 13-15 compounds. For example, InSb, AlP, GaAs, etc. 12-16 group compounds : Combination of elements of groups 12 and 16 yield some solid

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Class 12 ...

Answer: (i) Those solids which have 3-D arrangement of atoms, ions or molecules are called crystalline solids, e.g. diamond.

(ii) This is due to shifting of an ion from its normal position to an interstitial, in the crystal lattice, thus causing a vacancy in

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State  
the original position. The overall density is not affected by this change.

Important Questions for CBSE Class 12  
Chemistry The Solid ...

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The Solid State Class 12 Important Questions Long Answer Type (LA)  
Question 106. (a) An element has an atomic mass  $93 \text{ g mol}^{-1}$  and density  $11.5 \text{ g cm}^{-3}$ . If the edge length of its unit cell is  $300 \text{ pm}$ , identify the type of unit cell. (b) Write any two differences between amorphous solids and crystalline solids.

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(Delhi) (2017) Answer: (a) Given:

Important Questions for Class 12

Chemistry Chapter 1 The ...

Class 12 Important Questions for

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important resource for students preparing



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