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react with 10.4 g  $N_2$ ,  
producing 2.93 g  $NH_3$ .  
What is the theoretical  
yield in grams for this  
reaction under the given  
conditions? 1 day ago  
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true or false for each of  
the following  
statements. 1. In its

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sure to answer all parts.

Balance the following  
redox reaction and



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classify it as a  
combination,  
decomposition, or  
displacement reaction:

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7, s block

[Ar]3d<sup>10</sup>4s<sup>2</sup>4p<sup>3</sup> group  
15, period 4, p block

[Kr]4d<sup>10</sup>5s<sup>2</sup>5p<sup>5</sup> group  
17, period 5, p block

[Ne]3s<sup>2</sup>3p<sup>6</sup> group 18,  
period 3, p block

[Xe]6s<sup>1</sup> group 1, period

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6, s block [Ar]3d104s2  
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Homework. What is the mass of one atom of calcium.... How many

atoms are contained in

4.1 g o.... Choose from

the conversion factors

give.... LP gas (liquid

propane, C<sub>3</sub>H<sub>8</sub>) is the

fu....  $1.99 \times 10^{-23} \text{g}$ .

$2.1 \times 10^{23}$  atoms.

MASS (g) of Substance

A - Molar mass (g/mol)

of substance A... MA....

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