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IUPAC | NIE ~~Chapter 11 Alcohols And Ethers~~

11.27. Preparation of ethers by acid dehydration of secondary or tertiary alcohols is not a suitable method. Give reason. Ans: Acid catalysed dehydration of primary alcohols to ethers occurs by S_N2 reaction involving nucleophilic attack by

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the alcohol molecule on the protonated alcohol molecule.

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Kerala Plus Two Chemistry Notes Chapter 11 Alcohols, Phenols and Ethers. Alcohols and phenols are formed when a hydrogen atom in hydrocarbon is replaced by -OH group.

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Chapter 11 - Alcohols, Phenols and Ethers
Exercise 325 Solution 1 Primary alcohol '
(i), (ii), (iii) Secondary alcohol ' (iv), (v)

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The formation of ethers by dehydration of alcohol is a bimolecular reaction (S_N2) involving the attack of an alcohol molecule on a protonated alcohol molecule. In the method, the alkyl group should be unhindered. In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination

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and ethers part 1~~

Chemistry Chapter-11: Alcohols, Phenols
and Ethers. Q 11.1 : Give IUPAC names
of the compounds given below : Solution :

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(i) 2, 2, 4-Trimethylpentan-3-ol (ii) 5-Ethylheptane-2, 4-diol (iii) Butane-2, 3-diol (iv) Propane-1, 2, 3-triol (v) 2-Methylphenol (vi) 4-Methylphenol (vii) 2, 5-Dimethylphenol

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Intext Exercises. Question 1. Classify the

following as primary, secondary and

tertiary alcohols : Answer: (i) 1° , (ii) 1° ,

(iii) 1° , (iv) 2° , (v) 2° , (vi) 3° . Question 2.

Identify allylic alcohols in the above

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In cleavage of ethers, higher alcohol and lower alkyl halide is formed except in tert. butyl methyl ether, in which $(\text{CH}_3)_3\text{C}^+$ is more stable so $(\text{CH}_3)_3\text{C-I}$ and CH_3

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OH is formed. Anisole with HI gives phenol and CH_3I because phenoxide ion is stabilised by resonance.

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Alcohols, Phenols and Ethers Alcohols

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and Phenols Alcohols and phenols are formed when a hydrogen atom in hydrocarbon, aliphatic and aromatic respectively, is replaced by hydroxyl group (-OR group). Classification of Alcohols and Phenols In alcohols, -OR group is attached to Sp^3 hybridised carbon.

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Notes Chemistry Chapter 11. 1. Alcohols
and phenols may be classified as
monohydric, dihydric, trihydric or
polyhydric according to number of

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hydroxyl groups they contain one, two, three or many respectively in their molecules. 2.

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Alcohols and phenols are formed when a hydrogen atom in a hydrocarbon, aliphatic and aromatic respectively, is replaced by -OH group. The substitution of a hydrogen atom in a hydrocarbon by an alkoxy or aryloxy group (R-O/Ar-O) yields another class of compounds known as "ethers", for example, CH_3OCH_3 (dimethyl ether).

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should be unhindered. In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination dominates over substitution.

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Questions Chapter 11 Alcohols, Phenols
and Ethers Alcohols, Phenols and Ethers
Important Questions. Alcohols, Phenols
and Ethers Very Short Answer Type
Questions. Question 1. What is Ether?
Answer: Compounds formed by the
substitution of hydrogen atom of
hydrocarbon by alkoxy group are called

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Ethers. Question 2.

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